

Yale University

EliScholar – A Digital Platform for Scholarly Publishing at Yale

Yale Medicine Thesis Digital Library

School of Medicine

1-1-1845

Dissertation on medical chemistry

Joseph Edgar Clark
Yale University.

Follow this and additional works at: <https://elischolar.library.yale.edu/ymtdl>



Part of the [Medicine and Health Sciences Commons](#)

Recommended Citation

Clark, Joseph Edgar, "Dissertation on medical chemistry" (1845). *Yale Medicine Thesis Digital Library*. 3629.
<https://elischolar.library.yale.edu/ymtdl/3629>

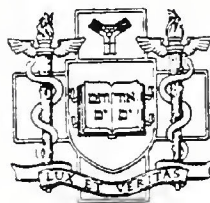
This Open Access Thesis is brought to you for free and open access by the School of Medicine at EliScholar – A Digital Platform for Scholarly Publishing at Yale. It has been accepted for inclusion in Yale Medicine Thesis Digital Library by an authorized administrator of EliScholar – A Digital Platform for Scholarly Publishing at Yale. For more information, please contact elischolar@yale.edu.



Digitized by the Internet Archive
in 2017 with funding from
Arcadia Fund

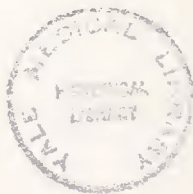
Harvey Cushing / John Hay Whitney
Medical Library

HISTORICAL LIBRARY



Yale University

Dissertations
read by the
Candidates for the Degree of Doctor in Medicine
at the
Annual Examination
in the
Medical Institution of Yale College,
January 15-16.
1845.



Archives
T113
Y11

IV.

Disputation
on
Medical Chemistry.

By
Joseph Edgar Clark,
of Rootstown, Ohio,
Candidate for the Degree of Doctor in Medicine.

Medical Chemistry -

According to the statutes of the state, it becomes necessary for candidates for degrees to present a dissertation, upon some subject, connected with the Medical Science; and I am well aware that something of this kind, will not only be expected, but absolutely required, and that excuses will avail nothing upon this occasion - I think however, it will not be expected, that the student of Medicine who is an attendant upon a rapid course of Lectures, in an Institution like this, which requires his strict and undivided attention to all the various branches that are continually urged upon his mind, and whose leisure moments for writing a dissertation are extremely few, and whose knowledge of Medicine, is merely theoretical - Under these circumstances therefore, I trust it will not be expected, that any thing new or original will be

written, or that I shall render myself in the slightest degree interesting, to the Honorable Members of the Medical Board, before which it is to be delivered; more especially as the subject which I have selected, is one with which every medical man, is already perfectly familiar. — In pursuing subject, I shall not confine myself strictly to what may be called Medical Chemistry, but deviate so far as to make a few general remarks relative to Chemistry as a science, its rise and progress, and its application not only in Pharmacy, and Physiology, but also in the arts, briefly glancing at various Chemical topics as they may occur to my mind, and as I can collect them from the writings of various authors, making in short my Thesis some what of a miscellaneous character —

Without ~~further~~ ceremony, I propose to proceed to my subject, feeling an assurance that all errors and mistakes will be overlooked

Of all the various branches of the Physical Sciences, there is perhaps none more highly interesting to the Scientific man, and of more extensive application in the arts, or of more practical importance to the Medical Profession in particular than Chemistry — Although to one unacquainted with the mysterious Laws by which it is governed, and the astonishing effects resulting from Chemical experiments &c, it would seem to be the least attractive, and most uninteresting, of all the subjects connected with the Sciences; in as much as it occupies itself, in the investigation of the composition, properties, & elementary Constitution ^{of all Substances}, and of the more obscure and apparently incomprehensible Phenomena, which are continually occurring in natural bodies many of which are unperceived by our senses, or unaccompanied by sensible motion.

But on the contrary; The present advanced state of perfection, to which the Chemical Science has already attained; the present amount

of Chemical knowledge, and facilities for obtaining it, the dexterity with which the Chemical Philosopher of the present day, in his Laboratory, manages to display or demonstrate Chemical facts, (of great importance to mankind), the brilliant and highly satisfactory results of his experiments, together with various other attractions all united cannot fail to render it decidedly the most interesting, and instructive branch of Natural Science —

Although it has been said that Chemistry as a science, could not be referred to a period, far beyond the commencement of the last century; yet we cannot for a moment doubt that the ancient Philosophers, attached great importance, to what was termed Chemistry or Alchemy, in as much, as it was the object of their researches, to discover ~~the~~ what was by them called, the Philosophers Stone, and with which they imagined

they should be enabled to Convert the baser metals into gold, & Silver, or to discover the Grand Catholicon, or universal remedy; which was to remove every disease, to avert death, and confer terrestrial immortality upon man; also the universal Solvent which it was supposed nothing could resist, and these Miraculous virtues resided in one and the same substance — Any farther than alchemy was concerned in the preparation or discovery of the Grand Catholicon or universal medicine which was fancied to be so admirably adapted for the removal of all diseases, and which was to prolong life to an indefinite length of time —

Any farther than this, it is generally believed, that Chemistry was not very extensively connected with Physic, until comparatively a late period —

How far alchemy had extended among the ancients or whether it had

even assumed the form of a sect, cannot be ascertained. Traces of it appear among the Greeks and Arabians - The Alchemists seem to have been established in the west of Europe, as early at least as the ninth Century. - Between the eleventh and fifteenth Centuries, it was in its most flourishing condition -

Cyber an Arabian Prince of the Seventh Century, is said to have been the earliest Alchemist whose name has reached posterity. - Also Rhases, & Avicenna, who were celebrated Physicians of the age, likewise Arabians, are said to be the first, who introduced Pharmaceutical preparations into their works; the latter of which, (Avicenna) describes particularly the method of conducting distillation. He also mentions for the first time the three mineral acids, and speaks likewise of sublimed arsenic, and corrosive sublimat.

In the year 1226, Roger Bacon laid the foundation of Chemical Science in Europe - His discoveries were so extraordinary,

that he was excommunicated by the Pope, & imprisoned for ten years, for supposed dealings with the Devil - He was it appears a believer in a universal Elixir for he proposed one to Pope Clement The tenth -

Basil Valentine, a German, who is said to have been born in 1344, led the way to the internal administration of metallic medicine by a variety of experiments on the nature of Antimony -

It is believed that preparations of Iron were administered internally, as the Sulphate of Iron, it being mentioned as a tonic styptic and astringent - He also mentions the Volatile Alkali, and its preparation from Sal. Ammoniac - and notices the production of Ether from Alcohol -

Paracelsus; who was born in the Year 1493 - is said to have used Mercury internally, although its effects externally & externally were well known in the twelfth Century -

Paracelsus, thus armed with opium⁸²
mercury and Antimony, remedies of notri-
fling importance, travelled in all directions,
and performing many extraordinary cu-
res, which brought him into notice, and
being well acquainted with Chemistry
Physic, was elected professor of Chemistry⁸³
in 1527 which was the first Professorship,
that was established, in Europe, for the pro-
motion and dissemination of Chemical Science.

While seated in his chair, he burnt with
great solemnity, the writings of Galen, &
Avicenna, and declared to his audience,
that if God would not impart the secre-
ets of Physic, it was not only allowable,
but even justifiable, to consult the Devil.

His contemporary Physicians he treated
with the utmost contempt, and told them
that the very down of his bald pate had
more knowledge, than all their writers,
the buckles of his shoes had more learning
than Galen & Avicenna, and his beard more

experience than all their universities — ⁹⁷

With such a temper and disposition however, it could not be expected, that he should long retain his Chair — He quitted it in consequence of a quarrel with the Magistrates — and although he boasted of possessing a Panacea, which was capable of curing all diseases in an instant, and even of prolonging life to an indefinite length — Yet this drunkard, & prince of Empirics, died after a few hours illness, with a bottle of his immortal Catholicon in his pocket —

About a century of Paracelsus, Van Helmont, a man of Eminence took the lead in Physic, He may be considered as the last of the Alchemists, his death completed the disgrace of the universal medicine.

He was the first Physician who applied calomel in uterine haemorrhage & acquired great reputation from the success of the practice —

The foundation of the Alchemistical system, being thus shaken the facts being which had been collected soon became a heap of rubbish, and Chemistry was left without any fixed principles, and destitute of an object —

About this period, or in the year 1669. Becher, & Stahl, published at Frankfurt a work entitled, *Physic Subteranea* — The publication of this book, forms a very important era, in the history of chemistry — It then escaped for ever from the trammels of Alchemy, and became the rudiments of the science which we find it at present. It has been distinguished by the name of the *Stahlian*, Theory —

Ever since the days of Stahl Chemistry has been cultivated with ardor in Germany, and the North, and the illustrious Philosophers, of these countries have contributed highly towards its progress, and rapid improvement —

11th

The most deservedly celebrated of these
are Margraf, Bergman, Scheele, Klaproth
&c. - In France soon after the establishment
of the Academy of Sciences in 1666 - Hom-
berg, Geoffroy, and Lemery, acquired great
celebrity by these chemical experiments
and discoveries; and after the new modelling
of the Academy, Chemistry became the
peculiar object of a part of that illustri-
ous body - Rouelle, who was made profes-
sor of Chemistry in Paris about the
year 1745 - Continued to infuse his own enth-
usiasm into the whole body of the French
literary men, and from that moment
Chemistry became the fashionable study -
Men of Eminence appeared every where;
discoveries multiplied, the spirit pervaded
the whole nation, extended itself over
Italy and appeared even in Spain -

After the death of Boyle,
and of some other of the earlier members of
the Royal Society little attention was paid

to Chemistry in Britain, except by a few individuals - The spirit which Newton had infused for the mathematical sciences, was so great, that for many years, they drew within their vortex, almost every man of eminence in Britain - But when Dr Cullen, became Professor of Chemistry in Edinburgh in 1756 - he kindled a flame of enthusiasm among the students, which was soon spread far and wide by the subsequent discoveries of Black, Cavendish, and Priestley and meeting with the kindled fires which were already burning in France, Germany, Sweden, and Italy, the Science of Chemistry burst forth at once, with unexampled lustre - Hence the rapid progress which it has made during the last fifty, the universal attention which it has excited, and the unexpected light, which it has thrown upon several of the most important arts and manufactures -

During all this time Chemistry has sustained amore or less intimate relation to the theory of medicine, The Pharmacopoeias and Dispensatories having kept pace with the scientific progress of the age - In tracing the march of chemical improvement, we cannot but be struck with the new, and powerful remedies which it has introduced; and the many unimportant and feeble articles, which it has dismissed from medical practice - Opium has been at length compelled, to confess its secret source of action, and Ipecacuan to yield its emetic element in a state of perfect purity - Chemistry has also been the means of establishing the identity of many bodies which were long considered specifically different; thus an extensive list of animal substances have been discarded since it is known that they owe there properties to one and the same common principle as to gelatine, albumen,

carbonate of lime, &c. so again the fixed ^{alk-}aline salt, produced by the incineration of different vegetables, have been found to be potash from whatever plant it may have been obtained with the exception of sea plants which yield soda.

For the foregoing sketch, or history, of the rise and progress of Chemistry thus far; I am indebted principally to the historical introduction, of Paris's Pharmacologia and the introductory remarks of Thomsons Chemistry, and others —

Some knowledge of Chemical arts, is said to be coeval with the earliest ages of antiquity — The application of Chemistry in the arts at the present time is very extensive, and it is more or less intimately connected with almost every branch of human industry — It is by the aid of Chemistry that, almost every kind of manufacturers are enabled successfully to go through with many of their most delicate processes, as bleaching, dye-

ing, and painting, also the manufacture of glass and the various kinds of pottery, and porcelain - It likewise furnishes mankind with a great variety of articles quite indispensable in domestic life, as sugar, vinegar, common salt, Sal Exalt's, Soap, coloring materials &c. - So also, in the more refined arts and acquirements as in daguerreotype drawing, plating by means of Galvanism, & the preparations of materials for painting -

Again it has been found, by comparatively recent investigations and experiments, that chemistry bears an intimate relation, to agriculture, and this subject is now exciting the attention of the agriculturist, and is cultivated with great interest by men of science and learning, and anyone who has examined the works of Liebig and others, on the subject of agricultural Chemistry, will not hesitate to say with me that Chemistry contributes not a little, to the improvement, & advancement, of that branch

16

of human industry, most of all to be cherished
among enlightened and civilized people -

Chemistry as a science -

Chemistry is defined by
Dr Thomson to be "That" Science which treats
of those events or changes in natural bodies
which are not accompanied by sensible
motions" - "Chemical events are equally num-
erous" says the same Author, "and fully as impor-
tant, as those, which belong to mechanical or Nat-
ural Philosophy for the Science comprehends un-
der it almost all the changes in natural
objects with which we are more immediately
connected, and in which we have the greatest
interest - Chemistry therefore is highly worthy
of our attention, not merely for its own sake,
because it increases our knowledge and
gives us the noblest display of the wisdom
and goodness of the Author of ~~the~~ Nature - But
it adds to our resources by extending our dom-
inions over the material world, and is therefore
calculated to promote our enjoyment and amuse-

17

ment our power - Again he remarks that Chemistry is intimately ^{connected} with a great variety of natural phenomena - All satisfactory explanation, of the causes of rain, snow, hail, dew, wind, earthquakes, & volcanoes, have been given by the aid of chemical knowledge - The phenomena of respiration, the decay and growth of plants, and some of the most important functions of the animal economy, are also explained in a satisfactory manner, only by the aid of Chemistry - The laws which govern chemical affinity, the doctrine of definite proportions, the chemical effects of electricity and galvanism the expansive power of heat evaporation the phenomena of flame and of combustion, are among some of the most prominent which arrest the attention of the chemical student - The doctrine of definite proportions, or atomic numbers, is not only highly interesting as a subject of Philosophy, but is also intimately connected with Chemistry, both as a science, and a practical

18-

art. Nothing perhaps, ^{even} the sublimest works
of nature, are not more calculated to elicit
the wonder and admiration of a reflecting ^{mind} than
this fact that substances combine with each
other in exact and definite quantities, or prop-
ortions and that these proportions are the same in
relation to the same substance through out the
world, and have been so ever since the creation.
This discovery may be considered as a new
proof, directed expressly, to the present age,
that the most minute works of what we call nature
do indeed bear the most indubitable marks of
divine agency and design - But while the discov-
ery itself, is an evidence of the profound Philosophy
of the present age, the development of its prin-
ciples by the constant accession of new ideas, is
calculated, rather to humble the pride of human
knowledge by as constant conviction that aft-
er all our acquisitions we know comparativ-
ly nothing of the laws and operations of nature.
The very fact that the laws of proportions
now, comparatively just known; to man, have existed

ever since the creation of matter, and have been in perpetual exercise all over the universe without a suspicion of their existence, is of itself a sufficient proof of the almost entire ignorance of man - even of the phenomena of nature, and a still stronger proof of his ignorance of her laws. "And if facts in themselves so simple, yet so wonderful, and when once known so obvious, have escaped the observation of man for thousands of years, is it not probable that phenomena are constantly going on before our eyes, which could we understand them, would astonish us still more - and at the same time, afford a still stronger proof of our ignorance, and want of penetration?"

In the next place, I come briefly to consider the relation which chemistry bears to Physiology, and more particularly that branch which relates to the nutrition and growth of the animal system, and the production of animal heat &c. I am very well aware that this is a subject upon which there is a great difference of opinion, and that there are many who are unwilling to admit that Chemistry has any thing

20
what ever to do with Physiology, that it is a
vital process only which supplies animal heat &c.

In pursuing this subject, I shall copy largely
from Liebig's writings, - although there are some
objections to his theory, yet it cannot be denied
that he advances many very ingenious and satis-
factory arguments in support of his doctrine -

In the preface to his work on organic Chemistry,
in its application to Physiology & Pathology, the
above named Author, says that "before the time
of Lavoisier's Scheele & Priestly Chemistry was not
more closely related to Physics than she is
now to Physiology. At the present day it is
so fused as it were into Physics that it would
be a difficult matter to draw the line between
them distinctly, - The connection between Chemistry
and Physiology is the same and in a nother half
Century it will be found impossible to separate
them;" he furthermore remarks that "one of the
most remarkable effects of the recent progress
of science is the alliance of Chemistry with
Physiology, by which a new and unexpected light

has been thrown upon the vital processes of plants and animals," "We have now no longer any difficulty in understanding the different actions of aliments, poisons and remedial agents, we have a clear conception of the causes of hunger, of the exact nature of death, and we are not as formerly obliged to content ourselves with a mere description of these symptoms."

"It is now ascertained with positive certainty that all the substances which constitute the food of man must be divided into two great classes one of which serves for the nutrition and reproduction of the animal body, while the other ministers to quite different purposes"—

"Thus Starch, Gum, Sugar, beer, wine, Spirits, &c furnish no element capable of entering into the composition of the blood muscular fibre, or any part which is the seat of the vital principle. The primary conditions of the maintenance of animal life, are a constant supply of certain matters animal food, and of oxygen in the shape of atmospheric air—

22

During every moment of life, oxygen is absorbed from the atmosphere into the organs of respiration, & the act of breathing cannot cease while life continues. The observations of Physiologists have demonstrated that the body of an adult man, supplied abundantly with food, neither increases nor diminishes, in weight, during twenty-four hours, and yet the quantity of oxygen absorbed into the system in that period, is very considerable.

According to the experiments of Lavoisier, an adult man takes into his system in one year not less than 746 pounds weight of oxygen, but we find his weight at the end of the year either exactly the same, or differing one way or the other by at most a few pounds. No part of the oxygen remains in the body but is given out again in combination with Carbon and Hydrogen - The Carbon and Hydrogen of certain parts of the animal body, combines with the oxygen introduced through the lungs and skin, and passes off in the form of carbonic acid and the vapor of water - We may assume with Lavoisier and Bergeus that an adult man absorbs into his system

32 1/4 oz of oxygen daily, and farther that the weight of the whole mass of blood, is 24 pounds, of which 80 per cent is water. Now from the known composition of the blood, we know that in order to convert its whole amount of carbon and hydrogen into carbonic acid and water, 64, 102, grs. oxygen are required, this quantity, will be taken into the system in four days and 5 hours. It follows then that there must be a sufficient quantity of carbon & hydrogen taken into the system, in the form of food, to supply the oxygen. And unless this is the case, there will be a diminution in the weight, the oxygen combining with the tissues of the body containing these elements, which will be exhaled from the lungs and skin in the form of carbonic acid and water. Hence it is evident that the amount of nourishment required by an animal for its support must be in a direct ratio with the quantity of oxygen taken into the system. The consumption of oxygen in a given time, may be expressed by the number of respirations. It is therefore obvious that in the same animal the quantity of nourishment required must vary with the force and number of respirations.

A child breathes quicker than an adult, and consequently requires food more frequently, and proportionably, in larger quantities, and bears hunger less easily - A bird, deprived of food dies on the third, while a serpent will live three months or longer without food - Air is expanded by heat, and ~~and~~ contracted by cold. An equal volume of hot and cold air contains therefore an unequal amount of oxygen - we expire more carbon at a low, than at a high temperature, and require more or less carbon in our food in the same proportion, and consequently more is respired in the frigid, than in the torrid zone - Even if an equal weight of food is consumed, in hot and cold climates, Infinite Wisdom has ordered that very unequal proportions of carbon shall be taken in it - as the food prepared by the inhabitants of Southern Climates, do not contain more than 12 per cent of Carbon, while the blubber and train oil which feed the inhabitants of the polar regions contain from 66 to 80 per cent of that Element -

From what has been said we may conclude that the oxygen of the atmosphere, received into the blood in the Lungs, and circulating through out every part of the animal body, acting on the elements of the food, is the source of animal heat.

All living creatures whose existence depends upon the absorption of oxygen, possess within themselves, a source of heat independent of surrounding objects —

In the animal body heat is produced only in those parts, ~~not~~ which arterial ~~arterial~~ blood, and with it the oxygen absorbed in respiration is conveyed — The combination of a combustible substance with oxygen is under all circumstances the only source of animal heat —

In whatever way carbon may combine with oxygen, the act of combination is accompanied by the disengagement of heat —

It is indifferent whether this combination takes place rapidly, or slowly, at a high, or at a low temperature, the amount of heat is

26

a constant quantity - The Carbon of the food being converted into carbonic acid within the body must give out exactly as much heat as if it had been directly burnt in oxygen gas, or common air, the only difference is the production of the heat, is diffused over unequal times - In oxygen gas the combustion of carbon is rapid and the heat is intense - in atmospheric air it burns slower and for a longer time - in the animal body the combination is still more gradual and the heat is lower in proportion - In the animal body the food is the fuel - In the winter when we take exercise in a cold atmosphere, and when consequently the amount of inspired oxygen, increases; the necessity for food containing Carbon & hydrogen increases in the same ratio, and by gratifying the ^{arise} thus excited, we obtain the most efficient protection against the most piercing cold, a starving man is soon frozen to death - The animals of prey in the arctic region as every one knows far exceed in voracity those of the torrid zone, our clothing is nearly an equivalent

for a certain amount of food - If we were to go naked, like certain savage tribes, or if in hunting or fishing we were exposed to the same degree of cold as the Samoyedes we should be able with ease to consume 10 pounds of flesh, and perhaps a dozen tallow candles into the bargain daily, as warmly clad travelers have related with astonishment of these people.

According to the preceding exposition the quantity of food is regulated by the number of respiration, by the temperature of the air, and by the amount of heat given off to the surrounding medium - The whole process of respiration appears most clearly developed when we consider the state of a man or other animals, totally deprived of food - The first effect of starvation, is the disappearance of the fat, and this fat cannot be traced either in the urine or in the scanty feces - Its carbon & hydrogen has been given out through the skin and lungs, in the form of oxidized products, it is obvious that they have served to support respiration - In the progress of starvation however it is not only the fat, which disappears, but also by degrees

all such of the solids, as are capable of being resolved as muscles and other of the organized tissues, and towards the end the particles of the Brain begin to undergo the process of oxidation, and delirium mania, and death closes the scene - That is to say all the resistance to the oxidizing power of the atmospheric oxygen ceases, and the chemical process of exsiccation or decay commences, in which every part of the body, the bones excepted, enters into combination with oxygen - The time which is required to cause death by starvation, depends on the amount of fat in the body, on the degree of exercise, as in labor, or exertion of any kind, on temperature of the air, and finally on the presence or absence of water - In all chronic diseases death is produced in the same way viz: the chemical action of the atmosphere, overcoming the vital action or force -

Again Liebig says that the substance of which the body of man is composed may be divided into two classes, into nitrogenous and nonnitrogenous, the former are capable of

Conversion into blood, the latter are incapable of this transformation - Out of those substances which are adapted to the formation of blood are formed all the organic tissues - The other class of substances serve to support the process of respiration -

Among the former are reckoned vegetable fibrine, Albumen, & caseine - Animal flesh & blood or animal fibrin, albumen, and caseine - Among the elements of respiration in our food, are fat, starch, gum, sugar, pectine, &c - The most recent and exact researches have established as a universal fact, to which nothing yet known is opposed that the nitrogenized constituents of vegetable food have a composition identical with that of the constituents of the blood - No nitrogenized compound, the composition of which differs from that from that of fibrine albumen and caseine is capable of supporting the vital process in animals - Chemical analysis has led to the remarkable result, that fibrine and albumen either vegetable or animal - and likewise caseine that nitrogenized constituent of milk, (the only one

upon which the growth of the organized tissues ~~stands~~ of the young animal depends, contain the same organic elements united in the same proportions, that is that they are isomeric, in their chemical composition, the proportion of these ultimate elements being identical, but the difference of their external properties shows that the particles of which they are composed are arranged in a different order. Both fibrine and albumen in the process of nutrition, are capable of being converted into muscular fiber, and muscular fiber is capable of being converted into blood. If now we compare the composition of organized parts with that of fibrine and albumen the following relations present themselves. All parts of the animal body which have a decided shape, which forms parts of organs, contain Nitrogen. No part of any organ which possesses motion and life is destitute of nitrogen, all of them contain likewise Carbon & the elements of water. Finally water & common salt are those ingredients of the body which are destitute of nitrogen. Both are amorphous or unorganized, and only so far take part in the vital process, as that their presence

34

is required for the due performance of the vital functions - Much more might be written upon this subject, but time and space will not permit, neither does the occasion require it -

We come next to the subject of medical Chemistry; or that branch of the chemical science which aids in the preparation of many of our most efficacious, and powerful medical remedies - If we were to draw a line of distinction between those medical substances which are formed by a purely chemical process and those which are not; we should find that a large majority of our most potent remedies would be included under the chemical division -

Doct: A T Thomson, says that "the operations of Pharmacy which are strictly chemical may be arranged in three classes -

First, operations which produce changes separating the constituents without any obvious decomposition -

Second, operations in which changes are produced by the chemical action of one set of bodies

upon a mother, or attended with obvious decomposition

Third, Operations in which the oxidizement and the deoxidizement of bodies are effected by means of a very high Temperature —

In the first Operations, the Changes are effected, 1st, By Caloric, inducing Liquefaction, Fusion, Evaporation, Distillation, Sublimation, &c. and 2^d, By water, and other fluids, as Solution, Maceration, Infusion Decoction &c.

In the Second Operations, the Changes are by, Decomposition, Dissolution, Precipitation, Crystallization & Fermentation —

The Third, and last Class of operations in which oxidizement and deoxidizement are effected by means of a high temperature as fusion, Distillation, Sublimation, the oxidizement of metals, The Deoxidizement or reduction of metals

Thus we find the above classification embracing an almost endless variety of medical bodies, as for example the Alkalies, Potassa, Soda, and Ammonia — All the Metallic Oxides, as of Silver Mercury Copper Iron Lead Zinc

Antimony & Arsenic - So also all the acids as Sulphuric, Carbonic, Boracic, Acetic, Benzoic, and so on - And a long list of precipitates, as the neutral Salts, and numerous others widely differing in their sensible properties -

It was my intention to have dwelt longer upon this part of my subject, but, ^{went up} time, and other circumstances ^{will} ~~prevent~~ not permit - and furthermore I am well aware that quite enough has already been written unless better justice can be done the subject,

Joseph E. Clark



YALE MEDICAL LIBRARY



3 9002 08670 4799

Accession no. 22994

Author

Yale Univ.

Theses for... M.D.
1845

Call no.

T113

Archives

Y11

